# metal-organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 294 KMean  $\sigma(\text{C-C}) = 0.005 \text{ Å}$  R factor = 0.032 wR factor = 0.076 Data-to-parameter ratio = 17.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# Chloro(2-{[N-(4-methoxybenzyl)methylamino]methyl}ferrocene- $\kappa^2 N, C^1$ )(triphenylphosphine- $\kappa P$ )palladium(II)

In the title compound,  $[PdCl{Fe(C_5H_5)(C_{15}H_{17}NO)}]$ -(C<sub>18</sub>H<sub>15</sub>P)] or  $[FePd(C_5H_5)(C_{15}H_{17}NO)Cl(C_{18}H_{15}P)]$ , the Pd<sup>II</sup> atom is in a slightly distorted square-planar environment. The dihedral angle between the two cyclopentadienyl rings of the ferrocenyl group is 6.4 (1)°.

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# Comment

Cyclopalladation of *N*-donor ligands, especially those bearing a ferrocenylimino group, have been extensively studied due to their applications in organic synthesis such as in the Heck reaction (Iyer & Ramesh, 2000) and the Suzuki coupling reaction (Weissmann & Milstein, 1999), *etc.* As a part of our ongoing investigations of cyclometallation of *N*-methyl-*N*ferrocenylmethylbenzyamines (Wang *et al.*, 2006), a new compound, (I), has been prepared and we report its crystal structure here.



In the compound (I), atom Pd1 is in a slightly distorted square-planar environment (Fig. 1 and Table 1). Atoms Pd1, P1, N1, Cl1 and C6 deviate from the mean plane through them by 0.0096 (7), -0.1110 (9), -0.1251 (10), 0.0963 (9), 0.1303 (11) Å, respectively. The dihedral angle between the two cyclopentadienyl rings of the ferrocenyl group is 6.4 (1)°. The substituted cyclopentadienyl plane forms a dihedral angle of 82.3 (1)° with the C14–C19 benzene ring. Except for an intramolecular C–H···Cl interaction (Table 2), no hydrogen bonds are observed in the crystal structure.

## **Experimental**

A solution of sodium tetrachloropalladate(II) (290 mg, 1 mmol) in methanol (15 ml) was added dropwise to a stirred solution of {[(N-methyl-N-4-methoxybenzyl)amino]methyl}ferrocene (350 mg, 1 mmol) and sodium acetate (82 mg, 1 mmol) in methanol (30 ml). The mixture was stirred at room temperature for 4 h. Then triphenylphosphine (410 mg, 1.5 mmol) was added and the mixture was stirred for another 30 min. The solvent was removed *in vacuo* and the residue was purified by column chromatography (silica gel, eluant: ethyl acetate / petroleum ether (333–363 K), 1:3) to give compound (I) (Yield: 80%). Analysis calculated for C<sub>38</sub>H<sub>37</sub>ClFe-NOPPd: C 60.66, H 4.96, N 1.86%; found: C 60.38, H 4.89, N 1.82%.

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## Crystal data

 $[FePd(C_{3}H_{5})(C_{15}H_{17}NO)-Cl(C_{18}H_{15}P)]$   $M_{r} = 752.36$ Monoclinic,  $P2_{1}/c$  a = 17.6945 (19) Å b = 16.1753 (17) Å c = 12.1585 (13) Å  $\beta = 106.128$  (2)° V = 3343.0 (6) Å<sup>3</sup> Z = 4

#### Data collection

Bruker APEX-II CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{min} = 0.670, T_{max} = 0.798$ 18615 measured reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.032$   $wR(F^2) = 0.076$  S = 1.026828 reflections 399 parameters H-atom parameters constrained

### Table 1

Selected geometric parameters (Å, °).

Pd1-C6	1.995 (3)	N1-C12	1.484 (3)
Pd1-N1	2.184 (2)	N1-C13	1.493 (3)
Pd1-P1	2.2264 (8)	N1-C11	1.499 (4)
Pd1-Cl1	2.3889 (8)		
C6-Pd1-N1	82.70 (10)	C6-Pd1-Cl1	171.93 (8)
C6-Pd1-P1	91.36 (8)	N1-Pd1-Cl1	91.80 (7)
N1-Pd1-P1	170.80 (6)	P1-Pd1-Cl1	94.79 (3)

 $D_x = 1.495 \text{ Mg m}^{-3}$ Mo *K* $\alpha$  radiation

reflections

 $\theta = 2.4-26.3^{\circ}$  $\mu = 1.13 \text{ mm}^{-1}$ 

T = 294 (2) K

Prism, red

 $R_{\rm int} = 0.036$ 

 $\theta_{\rm max} = 26.4^{\circ}$ 

 $h = -22 \rightarrow 20$ 

 $k = -9 \rightarrow 20$ 

 $l = -14 \rightarrow 15$ 

+ 0.4688P]

 $(\Delta/\sigma)_{\rm max} = 0.001$ 

 $\Delta \rho_{\rm max} = 0.37 \ {\rm e} \ {\rm \AA}^{-3}$ 

 $\Delta \rho_{\rm min} = -0.48 \text{ e } \text{\AA}^{-3}$ 

Cell parameters from 6333

 $0.30 \times 0.26 \times 0.20 \mbox{ mm}$ 

6828 independent reflections

5059 reflections with  $I > 2\sigma(I)$ 

 $w = 1/[\sigma^2(F_o^2) + (0.0342P)^2]$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

Tab	le	2
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Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$C13-H13A\cdots Cl1$	0.97	2.75	3.387 (3)	124

All H atoms were initially located in a difference Fourier map. The methyl H atoms were then constrained to an ideal geometry with C—H distances of 0.96 Å and  $U_{iso}(H) = 1.5U_{eq}(C)$ , but each group was allowed to rotate freely about its C—C bond. All other H atoms were placed in geometrically idealized positions and constrained to ride on



#### Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms are represented by circles of arbitrary size.

their parent atoms, with C–H distances in the range 0.93–0.97 Å and  $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C})$ . The  $U^{ij}$  components of the atoms C23 and C24 were restrained to approximately isotropic behaviour.

Data collection: *SMART* (Bruker, 2003); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1998); software used to prepare material for publication: *SHELXTL*.

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